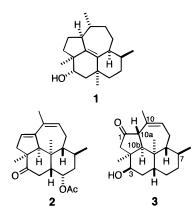
Synthetic Studies toward the Kempane Diterpenes: Preparation of the Ring System

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Received May 29, 1997

The defensive secretions of nasute soldier termites include tetracyclic diterpenes of two closely related types, the rippertanes (e.g., 1) and the kempanes (e.g., 2 and 3).¹ We report herein an approach to the synthesis of the ring system of the kempane diterpenes in a manner that should allow access to all of the known kempanes from one multifunctional pentacyclic compound. Previous approaches to the kempanes have led to only one total synthesis. The Dauben group² prepared kempene-2 (2) by a clever route in which, at different points in the synthesis, two rings were established by Diels-Alder additions of isoprene. Paquette and co-workers³ reported an approach to kempane **3** that included a remarkably efficient palladium-catalyzed [3 + 2] cycloaddition. An isomer of 3 was prepared in which the double bond was conjugated with the ketone, but, unfortunately, they were unable to transform this isomer into the less-stable⁴ natural product. In the only enantioselective approach, Metz et al.⁵ assembled a nor-methyl skeleton for 3α -hydroxy-15rippertene (1) from (-)- α -santonin.



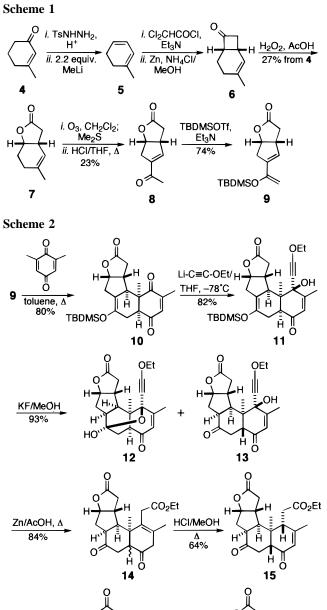
Our strategy was to use a Diels-Alder reaction to establish the relative stereochemistry for the central stereogenic centers at C-10a, C-10b, and C-10c.⁶ The diene was required to have oxygens at the same locations as in **2** and/or **3**, and functionality should already be present that would be needed for the closure of the final, seven-membered ring. An appropriate diene (**9**) was constructed from enone **4** by a route reminiscent of Corey's work,^{7,8} as outlined in Scheme 1. The trapping of dichloroketene by diene **5** proceeded regiospecifically,⁹ as did a

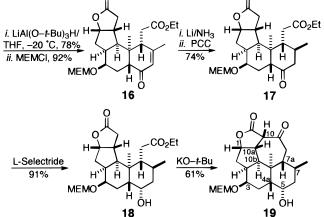
- (1) (a) Baker, R.; Walmsley, S. *Tetrahedron* **1982**, *38*, 1899–1910. (b) Prestwich, G. D. *Tetrahedron* **1982**, *38*, 1911–1919.
- (2) Dauben, W. G.; Farkas, I.; Bridon, D. P.; Chuang, C.-P.; Henegar, K. E. J. Am. Chem. Soc. **1991**, *113*, 5883–5884.
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 - (4) Taber, D. F. Tetrahedron Lett. 1993, 34, 1883-1884.

(5) Metz, P.; Bertels, S.; Fröhlich, R. J. Am. Chem. Soc. 1993, 115, 12595-12596.

(6) The numbering of kempane **3** is used for all of the compounds shown in Scheme 2.

(8) A lactone very similar to 7 has been resolved with (+)-1-(1-naph-thyl)ethylamine: Corey, E. J.; Snider, B. B. J. Org. Chem. **1974**, *39*, 256–258.





subsequent Baeyer–Villiger reaction ($6 \rightarrow 7$). Ring scission of 7 and acid-mediated aldol gave enone 8, which was most efficiently converted to the silyl enol ether 9 using the triflate and triethylamine.¹⁰

⁽⁷⁾ For instance, see: Corey, E. J.; Ravindranathan, T. *Tetrahedron Lett.* **1971**, 4753–4755.

⁽⁹⁾ Chemo- and regiospecific trapping of ketene with a 1-substituted 1,3-cyclohexadiene can also be seen in: Harding, K. E.; Strickland, J. B.; Pommerville, J. J. Org. Chem. 1988, 53, 4877–4883.
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Communications to the Editor

As shown in Scheme 2,⁶ the Diels-Alder reaction of 9 with 2,6-dimethyl-1,4-benzoquinone proceeded smoothly in a regio-, stereo-, and facially-selective manner to provide adduct 10, in which three of the four rings of the kempanes were already in place. An examination of the relative accessibilities¹¹ of the two ketones in 10 to axial attack led to the correct prediction of addition of ethoxyacetylide to provide 11 exclusively. Release of the silvl group with fluoride ion gave 12 and 13 in a 7:1 ratio, respectively. Formation of the hemiacetal 12 confirmed the relative stereochemistry at the carbinol center in 11. NOE measurements established that the minor product 13 had epimerized at C-4a. Reduction and solvolysis of the mixture of 12 and 13 was carried out in a single operation with zinc in refluxing acetic acid. This yielded the β , γ -unsaturated ketone 14, which included a minor component that was epimeric at C-4a (6:1 cis to trans ratio). Treatment of 14 with HCl in methanol completed the epimerization process, and the double bond was brought into conjugation concomitantly providing only one enone, **15**. NOE measurements established that the methyl at C-10c and C-4aH were now trans and that C-4aH and C-7aH were *cis*, i.e., the ester chain had assumed the required, thermodynamically-preferred equatorial position. Reduction of 15 with NaBH₄ in 1:1 MeOH/CH₂Cl₂ at -78 °C¹² was completely chemoselective but not stereoselective. A 1.5:1 mixture, epimeric at C-3, was obtained, undesired equatorial alcohol was the major component. On the other hand, only

the ketone at C-3 was reduced, predominantly by equatorial addition of hydride, with LiAl(O-t-Bu)₃H,¹³ and the axial alcohol was protected with (2-methoxyethoxy)methyl chloride to give 16. The equatorial position for the C-7 methyl in 17 developed as a result of the dissolving-metal reduction of the enone in 16. Reduction of the remaining ketone by equatorial attack with L-Selectride provided 18. Dieckmann cyclization of the final seven-membered ring with KO-t-Bu afforded the pentacyclic target 19, which has all of the stereogenic centers of the kempane diterpenes in their correct configurations (once again confirmed by NOE measurements) and oxygen functionality on every ring. The oxygen functions at C-1, C-3, and C-5 are as found in various kempanes; the lactone can be reduced to a C-10 methyl or may be used to facilitate methylation; and a remaining methyl group (at C-2a) can be introduced by 1,4-addition to an α,β unsaturated C-1 ketone. Modification of 19 to 2 and 3 is currently underway in our laboratory.

Acknowledgment. We thank the Natural Sciences and Engineering Research Council of Canada for financial support.

Supporting Information Available: Procedures for the preparation of compounds **10–19** and ¹H and ¹³C NMR spectra of **10–19** (31 pages). See any current masthead page for ordering and Internet access instructions.

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⁽¹²⁾ Ward, D. E.; Rhee, C. K.; Zoghaib, W. M. Tetrahedron Lett. 1988, 29, 517–520.

JA971769N

⁽¹³⁾ LiAl(O-*t*-Bu)₃H provided the axial alcohol and its C-3 epimer in a 8:1 ratio and in a combined yield of 88%. The epimers were separable by flash chromatography. The relative stereochemistry at the carbinol center was assigned on the basis of NOE measurements.